

Year 9 Subject Name Medium Term Plan Autumn Term 1

Learning	Biology – 9A Genetics and Evolution	
Overview	Chemistry – 9E Making Materials	
(Contents)	Physics – 9I Forces and Motion	
Assessment	Assessments will occur at the end of each topic using exam style questions.	
Opportunities		
Textbooks	Exploring Science Int Y9 SB - 9781292294131 Pearson Science	
Published Lesson	on l	
Resources		

Home Learning	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/variation-differences-within-	
Resources	species/lessons?sid-cb730d=jDm-gm6d3b&sm=0&src=3	
	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/adaptations-competition-natural-selection-and-evolution/lessons?sid-1abfe3=bp8HRVvhaN&sm=0&src=3	
	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/heredity-and-dna/lessons?sid-dbb614=i9fGAjZLpU&sm=0&src=3	
	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/materials/lessons?sid- 422894=q_p-QueGQM&sm=0&src=3	
	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/climate-change-and-living-sustainably/lessons/waste-and-recycling?sid-2c74d6=_l9ETiEJXp&sm=0&src=4	
	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/moving-by-force/lessons?sid-62d450=WOHKqw0_BU&sm=0&src=3	
Knowledge	Link to knowledge Organisers (look on the TES!)	
organisers		

Week	Learning Overview / objective (outlined above)	What should pupils know, understand and be able to do by the end of the
Number		week?
		(Use clear Success criteria)

1	Environmental Variation	 Identify different types of environmental variation and explain their causes Explain how environmental variation can cause problems with
		classification
	Inherited Variation	Identify different types of inherited variation
		Explain how sexual reproduction causes inherited variation
		Identify a normal distribution
	DNA	Outline how the structure of DNA was discovered
		Explain the importance of DNA Results the relative by the second of DNA Results the relative by the relative by the second of DNA Results the relative by the relative by the second of DNA Results the relative by the relative b
		 Describe the relationship between chromosomes, DNA, genes, genetic information and nuclei
2	Genes and Extinction	Explain how organisms become endangered or extinct
		Explain how adaptations affect the survival of organisms
		Explain some ways of preserving biodiversity
	Natural Selection	Recall that individuals in a population vary genetically
		Explain how natural selection works on these variations
	Investigation/science skills lesson	Carry out a practical/science skills lesson on the topic
	Consolidation Lesson	Completing content
3	End of Unit Assessment (Genetics and Evolution)	Apply knowledge of Genetics and Evolution to exam style questions
	About Ceramics	Name some examples of ceramics and their uses
		Explain why certain ceramics have particular uses
		Explain how the properties of ceramics can depend on their structure
	Polymers	Name some examples and uses of polymers
		Explain some of the main properties of polymers
		Describe how polymers are made
4	Composite Materials	Explain composite materials, giving examples
		Describe and justify the uses of some composite materials
		Explain what happens in thermal decomposition, and exothermic and
		endothermic reactions
	Problems with materials	Explain how making and using materials can cause problems
		Suggest ways of reducing these problems
	Recycling Materials	Explain the advantages of recycling

		Describe the recycling of some materials
	Investigation/science skills lesson	Carry out a practical/science skills lesson on the topic
5	Consolidation Lesson	Completing content
	End of Unit Assessment (Making Materials)	Apply knowledge of Making Materials to exam style questions
	Forces and Movement	 Recall the names of different types of forces Explain the effects of balanced and unbalanced forces Explain why moving objects have a top speed
6	Energy for Movement	 Recall ways in which energy can be stored and transferred Recall the law of conservation of energy State the meaning of efficiency
	Speed	 Describe the meanings of speed and mean (average) speed Use the formula relating speed, distance and time Represent simple journeys on a distance-time graph
	Turning Forces	 Describe how a simple lever can multiply forces or distances Identify the load, effort ad pivot on a diagram of a lever Describe the factors that affect the size of a moment Explain why something will balance if the moments are equal and opposite
	More Machines	 Describe how simple machines can magnify forces Describe the factors that affect the total work done
7	Investigation/science skills lesson Consolidation Lesson	Carry out a practical/science skills lesson on the topic Completing content
8	End of Unit Assessment (Forces and motion) Exam Skills Exam Skills Exam Skills	Apply knowledge of Forces and Motion to exam style questions
	Exam Skills	



Year 9 Subject Name Medium Term Plan Autumn Term 2

Learning	Biology – 9B Plant Growth	
Overview	Chemistry – 9F Reactivity	
(Contents)	Physics – 9J Force Fields and Electromagnets	
Assessment	Assessments will occur at the end of each topic using exam style questions.	
Opportunities		
Textbooks	Exploring Science Int Y9 SB - 9781292294131 Pearson Science	
Published Lesson		
Resources		

Home Learning	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/plant-nutrition-and-	
Resources	photosynthesis/lessons?sid-d859af=mXJxwYAD8l&sm=0&src=3	
	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/understanding-chemical-reactions/lessons?sid-93ea09=vrWDE2XEJA&sm=0&src=3 https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/series-circuits/lessons?sid-62c84a=DiWLHfEBbs&sm=0&src=3	
	https://www.thenational.academy/teachers/programmes/science-secondary-ks3/units/magnets-and-electromagnets/lessons?sid-b12abd=QP4h77gJJu&sm=0&src=3	
Knowledge	Link to knowledge Organisers (look on the TES!)	
organisers		

Week Number	Learning Overview / objective (outlined above)	What should pupils know, understand and be able to do by the end of the week? (Use clear Success criteria)
1	Reactions in Plants	Explain what happens when plants photosynthesise and respire
		Explain how the rate of photosynthesis can be affected
	Plant Adaptations	Describe how leaves, roots and stems are adapted for their functions
		Explain how substances enter and leave plants

	Plant Products	Explain how and why plants make different substances
		Explain the importance of nitrates
2	Growing Crops	Describe how pests and human populations alter the food supply
		Explain ways in which farmers boost food production
		Explain some ways in which plant varieties are created
	Farming Problems	Use models, for examples food webs and the carbon cycle, to explain
		changes in an ecosystem
		Recognise the advantages and disadvantages of different farming
		methods
	Investigation/science skills lesson	Carry out a practical/science skills lesson on the topic
	Consolidation Lesson	Completing content
3	End of Unit Assessment (9B Plant Growth)	Apply knowledge of Plant Growth to exam style questions
	Types of Explosion	Identify and explain the differences between physical changes and
		chemical reactions
		Model reactions using words and formulae
		Use particle theory to explain gas pressure and how it can be changed
	Reactivity	Describe the reactions of metals with water, dilute acids and air
		Explain how metals are placed in the reactivity series
		Describe the structure of an atom
4	Energy and Reactions	Describe the test for oxygen
		Explain how combustion reactions can be speeded up
		Classify changes as exothermic or endothermic
		Explain why some reactions need a supply of energy
	Displacement	Explain what happens in a displacement reaction
		Predict whether a displacement reaction will occur
	Extracting Metals	Explain why the method used to extract a metal is related to cost and
		the metal's reactivity
		Describe how metals are extracted from their ores by heating with
		carbon or by electrolysis
		Explain what happens in redox reactions
_	Investigation/science skills lesson	Carry out a practical/science skills lesson on the topic
5	Consolidation Lesson	Completing content
	End of Unit Assessment (9F Reactivity)	Apply knowledge of Reactivity to exam style questions
	Force Fields	State what is meant by a force field
		Describe the shape of a magnetic field

		Recall the factors that affect the strength of gravity
		Calculate the weight of a mass
6	Static Electricity	Explain why an insulating material can be given a charge by rubbing
		Describe how electrically charged objects affect each other
		Describe an electric field
	Current Electricity	Explain how switches can be used to control different parts of a circuit
		Recall how current behaves in series and parallel circuits
		 Describe how voltage behaves in series and parallel circuits
	Resistance	Describe some factors that affect resistance
		 Use the formula relating voltage, current and resistance
	Electromagnets	Describe an electromagnet and its magnetic field
		Describe how the strength of an electromagnet can be changed
		 Describe some applications of electromagnets
7	Investigation/science skills lesson	Carry out a practical/science skills lesson on the topic
	Consolidation Lesson	Completing content
	End of Unit Assessment (9J Force Fields and Electromagnets)	Apply knowledge of Reactivity to exam style questions



Year 9 Science Medium Term Plan Science Term 1

Learning	Overview of main learning objectives:	
Overview	Characteristic of living organisms	
(Contents)	Variety of living organisms	
	States of matter	
	Elements, compounds, mixtures	
Assessment	Assessments will occur at the end of each topic using exam style questions.	
Opportunities		
Textbooks		
Published Lesson		
Resources		
Home Learning		

Home Learning	
Resources	
Knowledge	Link to knowledge Organisers (look on the TES!)
organisers	

Week Number	Learning Overview / objective (outlined above)	What should pupils know, understand and be able to do by the end of the week? (Use clear Success criteria)
1	 Characteristics of living organisms Plants Animals 	 1.1 understand how living organisms share the following characteristics: they require nutrition they respire they excrete their waste they respond to their surroundings they move they control their internal conditions they reproduce
		they grow and develop.

		Plants: these are multicellular organisms; their cells contain chloroplasts and are able to carry out photosynthesis; their cells have cellulose cell walls; they store carbohydrates as starch or sucrose. Examples include flowering plants, such as a cereal (for example maize) and a nerbaceous legume (for example peas or beans). Animals: these are multicellular organisms; their cells do not contain chloroplasts and are not able to carry out photosynthesis; they have no cell walls; they usually have nervous coordination and are able to move from one place to another; they often store carbohydrate as glycogen. Examples include mammals (for example humans) and insects (for example nousefly and mosquito).
2	 Fungi Protoctists Bacteria Viruses 	Fungi: these are organisms that are not able to carry out photosynthesis; their body is usually organised into a mycelium made from thread-like structures called hyphae, which contain many nuclei; some examples are single-celled; their cells have walls made of chitin; they feed by extracellular secretion of digestive enzymes onto food material and absorption of the organic products; this is known as saprotrophic nutrition; they may store carbohydrate as glycogen. Examples include <i>Mucor</i> , which has the typical fungal hyphal structure, and yeast, which is single-celled. Protoctists: these are microscopic single-celled organisms. Some, like <i>Amoeba</i> , that live in pond water, have features like an animal cell, while others, like <i>Chlorella</i> , have chloroplasts and are more like plants. A pathogenic example is <i>Plasmodium</i> , responsible for causing malaria. 1.3 describe the common features shown by prokaryotic organisms such as bacteria Bacteria: these are microscopic single-celled organisms; they have a cell wall, cell membrane, cytoplasm and plasmids; they lack a nucleus but contain a circular chromosome of DNA; some bacteria can carry out photosynthesis but most feed off other living or dead organisms. Examples include <i>Lactobacillus bulgaricus</i> , a rod-shaped bacterium used in the production of yoghurt from milk, and <i>Pneumococcus</i> , a spherical bacterium that acts as the pathogen causing pneumonia. 1.4 understand the term 'pathogen' and know that pathogens may include fungi, bacteria, protoctists and viruses Viruses: these are not living organisms. They are small particles, smaller than bacteria; they are parasitic and can reproduce only inside living cells; they infect every type of living organism. They have a wide variety of shapes and sizes; they have no cellular structure but
		 1.3 describe the common features shown by prokaryotic of Bacteria: these are microscopic single-celled organisms cytoplasm and plasmids; they lack a nucleus but containsome bacteria can carry out photosynthesis but most of Examples include Lactobacillus bulgaricus, a rod-shaped yoghurt from milk, and Pneumococcus, a spherical bacticausing pneumonia. 1.4 understand the term 'pathogen' and know that pathogogorous protoctists and viruses Viruses: these are not living organisms. They are small are parasitic and can reproduce only inside living cells;

3	 Levels of organisation Cell structure and organelles Plant and animal cells 	2.1 describe the levels of organisation in organisms: organelles, cells, tissues, organs and systems 2.2 describe cell structures, including the nucleus, cytoplasm, cell membrane, cell wall, mitochondria, chloroplasts, ribosomes and vacuole 2.3 describe the functions of the nucleus, cytoplasm, cell membrane, cell wall, mitochondria, chloroplasts, ribosomes and vacuole 2.4 know the similarities and differences in the structure of plant and animal cells
4	 Consolidation Biology assessment States of matter and interconversion Elements, compounds, mixtures and pure substances 	1.8 understand how to classify a substance as an element, compound or mixture 1.9 understand that a pure substance has a fixed melting and boiling point, but that a mixture may melt or boil over a range of temperatures 1.1 understand the three states of matter in terms of the arrangement, movement and energy of the particles 1.2 understand the interconversions between the three states of matter in terms of: • the names of the interconversions • how they are achieved • the changes in arrangement, movement and energy of the particles

5	 Separation techniques – definitions Practical Practical Chromatography 	1.10 describe these experimental techniques for the separation of mixtures: • simple distillation • fractional distillation • filtration • crystallisation • paper chromatography 1.11 understand how a chromatogram provides information about the composition of a mixture
6	 Chromatography practical Consolidation Chemistry assessment 	1.12 understand how to use the calculation of R _f values to identify the components of a mixture 1.13 practical: investigate paper chromatography using inks/food colourings



Year 9 Science Medium Term Plan Spring Term 2

Learning	Atomic structure		
Overview	Periodic table		
(Contents)	Forces – movements and position		
,	Forces – movement and momentum		
Assessment	Assessments will occur at the end of each topic using exam style questions.		
Opportunities			
Textbooks	IG Biology SB 9780435185084 Pearson		
Published Lesson	77043310304 Tearson		
Resources	IG Chemistry SB	9780435185169	Pearson
	IG Physics SB	9780435185275	Pearson

Home Learning	https://www.thenational.academy/teachers/programmes/chemistry-secondary-ks4-higher-edexcel/units/atomic-	
Resources	structure-and-the-periodic-table/lessons	
	https://www.thenational.academy/teachers/programmes/physics-secondary-ks4-higher-edexcel/units/measuring-and-	
	<u>calculating-motion/lessons</u>	
	https://www.thenational.academy/teachers/programmes/physics-secondary-ks4-higher-edexcel/units/forces-make-things-	
	<u>change/lessons</u>	
Knowledge	Link to knowledge Organisers (look on the TES!)	
organisers		

Week Number	Learning Overview / objective (outlined above)	What should pupils know, understand and be able to do by the end of the week? (Use clear Success criteria)	
1	Atomic Structure	1.14 know what is meant by the terms 'atom' and 'molecule'	
		1.15 know the structure of an atom in terms of the positions, relative masses and relative charges of sub-atomic particles	
		1.16 know what is meant by the terms: atomic number, mass number, isotopes and relative atomic mass (A _r)	
		1.17 be able to calculate the relative atomic mass of an element (A _r) from isotopic abundances	

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	Periodic Table	1.18	understand how elements are arranged in the Periodic Table:
			in order of atomic number
		8	in groups and periods
		1.19	understand how to deduce the electronic configurations of the first 20 elements from their positions in the Periodic Table
		1.20	understand how to use electrical conductivity and the acid-base character of oxides to classify elements as metals or non-metals
		1.21	identify an element as a metal or a non-metal according to its position in the Periodic Table
		1.22	understand how the electronic configuration of a main group element is related to its position in the Periodic Table
		1.23	understand why elements in the same group of the Periodic Table have similar chemical properties
		1.24	understand why the noble gases (Group 0) do not readily react
2	Consolidation	1.3	plot and explain distance-time graphs
-	Chemistry assessment	1.4	know and use the relationship between average speed, distance moved and time taken:
	Distance time graphs		
	 Practical investigation collect data 		average speed = $\frac{\text{distance moved}}{\text{time taken}}$
	1 Tactical Investigation concet data	1.5	practical: investigate the motion of everyday objects such as toy cars or tennis balls
		1.6	know and use the relationship between acceleration, change in velocity and time taken:
			$acceleration = \frac{change in velocity}{time taken}$
			$a = \frac{(v - u)}{t}$
		1.7	plot and explain velocity-time graphs
		1.8	determine acceleration from the gradient of a velocity-time graph
		1.9	determine the distance travelled from the area between a velocity-time graph and the time axis
		1.10	use the relationship between final speed, initial speed, acceleration and distance moved:
			$(final speed)^2 = (initial speed)^2 + (2 \times acceleration \times distance moved)$
			$v^2 = u^2 + (2 \times a \times s)$

3	Write up of practical –distance time graphs	1.11 describe the effects of forces between bodies such as changes in speed, shape or direction
	Velocity time graphs	1.12 identify different types of force such as gravitational or electrostatic
	Introduction to forces	1.13 understand how vector quantities differ from scalar quantities
	 Forces, mass and acceleration - 	1.14 understand that force is a vector quantity
		1.15 calculate the resultant force of forces that act along a line
		1.16 know that friction is a force that opposes motion1.17 know and use the relationship between unbalanced force, mass and acceleration:
		force = mass × acceleration
		$F = m \times a$ 1.18 know and use the relationship between weight, mass and gravitational field strength:
		weight = mass × gravitational field strength
		$W = m \times g$
	1.21 describe the forces acting on falling objects (and explain why falling objects reach a terminal velocity)	
4	Weight, mass and terminal velocity Stanning and breaking distances.	1.22 practical: investigate how extension varies with applied force for helical springs, metal wires and rubber bands
	Stopping and breaking distancesHooke's law	1.23 know that the initial linear region of a force-extension graph is associated with Hooke's law
■ HOOKE S law	1.24 describe elastic behaviour as the ability of a material to recover its original shape after the forces causing deformation have been removed	
5	Hooke's Law practical	
	Hookes law write up	
	Consolidation lesson	
6	• Revision	
	 Physics assessment 	



Year 9 Science Medium Term Plan Summer Term 1

Learning	Overview of main learning objectives:			
Overview	Nutrition – flowering plants and humans			
(Contents)	Chemical formula, equations and calculations.			
	Mains electricity			
Assessment	Assessments will occur at the end of each topic using exam style questions.			
Opportunities				
Textbooks Published Lesson IG Biology SB 9780435185084				
Resources	IG Chemistry SB	9780435185169	Pearson	
	IG Physics SB 9780435185275 Pearson			
Home Learning	https://www.thenational.academy/teachers/programmes/biology-secondary-ks4-higher-edexcel/units/biological-			
Resources	molecules-and-enzymes/lessons			
https://www.thenational.academy/teachers/programmes/biology-secondary-ks4-higher-edexcel/units/phrogrammes/biology-second			ohotosynthesis-	
	https://www.thenational.academy/teachers/programmes/physics-secondary-ks4-higher-edexcel/units/mains-			

Home Learning	nttps://www.tnenational.academy/teachers/programmes/biology-secondary-ks4-nigher-edexcei/units/biological-	
Resources	molecules-and-enzymes/lessons	
	https://www.thenational.academy/teachers/programmes/biology-secondary-ks4-higher-edexcel/units/photosynthesis-	
	requirements-and-products/lessons	
	https://www.thenational.academy/teachers/programmes/physics-secondary-ks4-higher-edexcel/units/mains-	
	<u>electricity/lessons</u>	
	https://www.thenational.academy/teachers/programmes/physics-secondary-ks4-higher-edexcel/units/electric-fields-and-	
	<u>circuit-calculations/lessons</u>	
Knowledge	Link to knowledge Organisers (look on the TES!)	
organisers		

Week Number Learning Overview / objective (outlined above		What should pupils know, understand and be able to do by the end of the
	Learning Overview / objective (outlined above)	week?
		(Use clear Success criteria)

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1	PhotosynthesisInvestigating photosynthesis	2.18 understand the process of photosynthesis and its importance in the conversion of light energy to chemical energy
	• Leaf structure	2.19 know the word equation and the balanced chemical symbol equation for photosynthesis
		2.20 understand how varying carbon dioxide concentration, light intensity and temperature affect the rate of photosynthesis
		2.21 describe the structure of the leaf and explain how it is adapted for photosynthesis
		2.22 understand that plants require mineral ions for growth, and that magnesium ions are needed for chlorophyll and nitrate ions are needed for amino acids
		2.23 practical: investigate photosynthesis, showing the evolution of oxygen from a water plant, the production of starch and the requirements of light, carbon dioxide and chlorophyll
2	Balanced dietDigestive system	2.24 understand that a balanced diet should include appropriate proportions of carbohydrate, protein, lipid, vitamins, minerals, water and dietary fibre
	EnzymesBile and the small intestine	2.25 identify the sources and describe the functions of carbohydrate, protein, lipid (fats and oils), vitamins A, C and D, the mineral ions calcium and iron, water and dietary fibre as components of the diet
		2.26 understand how energy requirements vary with activity levels, age and pregnancy
		2.27 describe the structure and function of the human alimentary canal, including the mouth, oesophagus, stomach, small intestine (duodenum and ileum), large intestine (colon and rectum) and pancreas
		2.28 understand how food is moved through the gut by peristalsis
		2.29 understand the role of digestive enzymes, including the digestion of starch to glucose by amylase and maltase, the digestion of proteins to amino acids by proteases and the digestion of lipids to fatty acids and glycerol by lipases
		2.30 understand that bile is produced by the liver and stored in the gall bladder
		2.31 understand the role of bile in neutralising stomach acid and emulsifying lipids
		2.32 understand how the small intestine is adapted for absorption, including the structure of a villus
3	 Biology assessment 	1.25 write word equations and balanced chemical equations (including state symbols):
	 Word and symbol equations 	 for reactions studied in this specification
	 Relative formula mass 	for unfamiliar reactions where suitable information is provided
		1.26 calculate relative formula masses (including relative molecular masses) (M_r) from relative atomic masses (A_r)
4	• The mole	1.27 know that the mole (mol) is the unit for the amount of a substance
	Calculating reacting massCalculating percentage yield	1.28 understand how to carry out calculations involving amount of substance, relative atomic mass (A _r) and relative formula mass (M _r)
	Calculating percentage yieldEmpirical formulae	1.29 calculate reacting masses using experimental data and chemical equations
	- Empirical formulae	1.30 calculate percentage yield
		1.32 know what is meant by the terms 'empirical formula' and 'molecular formula'
		1.33 calculate empirical and molecular formulae from experimental data

5	 Practical – determining formulae Consolidation Chemistry assessment 	1.31 understand how the formulae of simple compounds can be obtained experimentally, including metal oxides, water and salts containing water of crystallisation 1.36 practical: know how to determine the formula of a metal oxide by combustion (e.g. magnesium oxide) or by reduction (e.g. copper(II) oxide)
• P • E	 Power = current x voltage Energy transferred, current, voltage and time. 	 2.2 understand how the use of insulation, double insulation, earthing, fuses and circuit breaker protects the device or user in a range of domestic appliances 2.3 understand why a current in a resistor results in the electrical transfer of energy and an increase in temperature, and how this can be used in a variety of domestic contexts 2.4 know and use the relationship between power, current and voltage: power = current × voltage P = I × V and apply the relationship to the selection of appropriate fuses
		 2.5 use the relationship between energy transferred, current, voltage and time: energy transferred = current × voltage × time E = I × V × t 2.6 know the difference between mains electricity being alternating current (a.c.) and direct current (d.c.) being supplied by a cell or battery